

ZAYTSEVA, M. I.: Master Phys-Math Sci (disc) ~ "Regulated groups". Moscow,  
1959. 9 pp (Min Education RSFSR, Moscow Oblast Pedagogical Inst im N. K.  
Krupskaya) (KL, No 11, 1959, 114)

ZAYTSEVA, M.G.

Respiration and mineral nutrition of plants. Fiziol. rast.  
12 no.5:794-804 S-0 '65. (MIRA 19:1)

1. Institut fiziologii rasteniy AN SSSR, Moskva.

ZAYTSEVA, M.G.; POZDNYAKOVA, V.A.

Some characteristics of soil nutrition in plants of the Far North.  
Dokl.AN SSSR 138 no.5;1223-1226 Je '61. (MIRA 14:6)

1. Institut fiziologii rasteniy im. K.A.Timiryazeva AN SSSR.  
Predstavлено академиком A.L.Kursanovym.  
(Murmansk Province—Wheat—Fertilizers and manures)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

ZAYTSEVA, N.G.; SHENKO, D.M.; TOVONYAKOVA, V.A.

Metabolism in wheat in connection with nutritional conditions in  
the Far North. Trudy lab. evol. i ekol. fiz. no.2:137-162 '67.  
(MIRA 18:3)

ZAYTSEVA, M.G.

Studies on nitrogen metabolism of wheat in the Far East. Fiziol.  
rast. 8 no.3:318-324 '61. (MIRA 14:5)

1. Institut fiziologii rasteniy im. K.A.Timiryazeva Akademii nauk  
SSSR, Moskva.  
(Murmansk Province--Wheat) (Nitrogen metabolism)

MAL'TSEV, Anatoliy Ivanovich; ZAYTSEVA, M.G., redaktor; GAVRILOV, S.S.,  
tekhnicheskiy redaktor

[Principles of linear algebra] Osnovy lineinoi algebry. Izd. 2-e,  
perer. Moskva, Gos. izd-vo tekhniko-teoret. lit-ry, 1956. 340 p.  
(Algebra) (MLRA 9:11)

ZAYKIN, M. G.

35961 O was born January 2, 1900, in Odintsovo, Moscow Obl., Russia. He is a doctor. He has been married since 1928. His wife's name is Nadezhda Ivanovna Zaykin.

EDUCATION: Higher School of Medicine, Moscow, Russia; Faculty of Internal Medicine, Moscow, Russia.

ZAYTSEVA, M.G.

Effect of low nighttime temperatures on growth and metabolism of barley plants  
in the early phases of their development. Izv.Otd.est.nauk AS Tadzh.SSR  
no.1:27-32 '52. (MLRA 9:10)

1.Institut botaniki Akademii nauk Tadzhikskey SSR. Pamirskaia biostantsiya.  
(Barley) (Plants--Metabolism)

ZAYTSEVA, M.G.

Effect of temperature on the germination of barley seeds. Izv.Otd.est.  
nauk AN Tadzh.SSR no.1:21-26 '52. (MLRA 9:10)

1.Institut botaniki Akademii nauk Tadzhikskoy SSR. Pamirskaya biesstan-  
tsiya.  
(Germination) (Barley)

GOLDOVSKIY, A.M., doktor tekhn.nauk; ZAYTSEVA, M.G., inzh.

Evaluation of the technological characteristics of sunflower seeds of  
the new improved varieties. Masl.-shir.prom. 29 no.1:7-11 Ja '63.  
(MIRA 16:2)

1. Vsesoyuznyy zaochnyy tekhnologicheskiy institut prikhchevoy  
promyshlennosti (for Goldovskiy). 2. Ukrainskiy nauchno-issledo-  
vatel'skiy institut maslozhirovoy promyshlennosti (for Zaytseva).  
(Sunflower seed--Testing)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

ZAITSEVA, T. G.

30/06/06 определен на земельный участок, расположенный в селе Борисоглебское, Наро-Фоминского района, Московской области, по адресу: д. Борисоглебское, ул. Мира, д. 11-17

SO: Летопись Народного Статса, №2, Москва, 1910

ZAYTSEVA, M.G.

KUZNETSOV, S.M.; ZAYTSEVA, M.G. [Zaitseva, M.H.]

Jubilee sessions of the institutes of the Academy of Sciences of the  
Ukrainian S.S.R. dedicated to the 40th anniversary of the Ukrainian  
Soviet Socialist Republic. Visnyk AN URSR 29 no.1:66-71 Ja '58.  
(MIRA 11:4)

(Ukraine--Science)

ZAYTSEVA, M.G.  
KUZNETSOV, S.M.; ZAYTSEVA, M.G. [Zaitseva, M.H.]

General jubilee session of the Academy of Sciences of the Ukrainian  
S.S.R. dedicated to the 40th anniversary of the Ukrainian Soviet  
Socialist Republic. Visnyk AN URSR 29 no.1:51-58 Ja '58.

(MIRA 11:4)

(Ukraine--Science)

ZAYTSEVA, M.G.; SEDENKO, D.M.; POZDNYAKOVA, V.A.

Effect of phosphorus and magnesium on the respiration of plants  
cultivated in the Kola Peninsula. Fiziol. rast. 9 no.1:98-105  
'62. (MIRA 15:3)

1. K.A.Timirazev Institute of Plant Physiology, U.S.S.R. Academy  
of Sciences, Moscow.

(Kola Peninsula--Plants--Respiration)

(Plants, Effect of phosphorus on)

(Plants, Effect of magnesium on)

GORDEN'EV, D.I., prof., glav. red.; DVORYANKIN, F.A., prof., red.;  
KOMONKOV, A.F., red.; RYBNIKOV, K.A., prof., red.; SOLOV'IEV,  
A.I., dotsent, red.; SPASSKIY, B.I., dotsent, red.; FIGUROV-  
SKIY, N.A., prof., red.; SHEVTSOV, N.S., prof., red.; KHIGIAN,  
A.Kh., prof., red.; ZAYTSEVA, M.G., red.; YERMAKOV, M.S., tekhn.  
red.

[History and methodology of the natural sciences] Istočnica i  
metodologija estestvennykh nauk. Moskva. No.1. [Physics] Fi-  
zika. 1960. 221 p. (MIRA 14:5)

1. Moscow. Universitet.  
(Physics)

PREDVODITELEV, A.S., prof.; MLODEZYEVSKIY, A.B., prof.; ZAYTSEVA, M.G.,  
red.; YERMAKOV, M.S., tekhn.red.

Ivan Filippovich Usagin. Moskva, 1959. 297 p. (MIRA 12:10)

1. Moscow. Universitet. Fizicheskiy fakul'tet. Kabinet istorii fiziki.
2. Chlen-korrespondent AN SSSR (for Predvoditelev).
3. Fizicheskiy fakul'tet Moskovskogo universiteta (for Mlózhevskiy).

(Usagin, Ivan Filippovich, 1855-1919)

ZAYTSEVA, M.G.

Investigating the process of nitrogen and phosphorus absorption  
by plant roots as affected by light conditions and temperatures  
at higher altitudes in the Pamirs. Trudy AN Tadzh. SSR 47:3-63  
'56. (MIRA 12:5)

(Pamirs--Plants--Assimilation)  
(Plants, Effect of light on)  
(Plants, Effect of temperature on)

LEVSHIN, Leonid Vadimovich; VOVCHENKO, G.D., prof., otv.red.; BERNSTEYN,  
S.B., prof., red.; VILENSKIY, D.G., prof., red.; GORDEYEV, D.I.,  
prof., red.; GUDZIY, N.K., prof., red.; ZAYONCHKOVSKIY, P.A., prof.,  
red.; KACHEK'IAN, S.F., prof., red.; MEL'NIKOVA, K.P., kand.nauk, red.;  
POLYANSKIY, F.Ya., prof., red.; RYBNIKOV, K.A., prof., red.; SKAZKIN,  
S.D., akademik, red.; SOLOV'YEV, A.N., dotsent, red.; ZAYTSEVA, M.G.,  
red.; GEORGIYEVA, G.I., tekhn.red.

Sergei Ivanovich Vavilov. Moskva, Izd-vo Mosk.univ., 1960. 101 p.  
(Zamechatel'nye uchenye Moskovskogo universiteta, no.24).

(MIRA 13:6)

(Vavilov, Sergei Ivanovich, 1891-1951)

KUZNETSOV, Ye.S., red.; ZAYTSEVA, M.G., red.; CHISTYAKOVA, K.S.,  
tekhn.red.

[Some mathematical problems on neutron physics] Nekotorye  
matematicheskie zadachi neitronnoi fiziki; sbornik rabot  
kafedry atomnogo iadra fizicheskogo fakul'teta MGU pod red.  
E.S.Kuznetsova. Moskva, Izd-vo Mosk.univ., 1960. 219 p.  
(Neutrons) (MIRA 13:7)

KONONKOV, Arkadiy Fedorovich; VOVCHENKO, G.D., prof., otv.red.; BERN-SHTEYN, S.B., prof., red.; VILENSKIY, D.G., prof., red.; GORIENEV, D.I., prof., red.; GUDZIY, H.K., prof., red.; ZAYON-CHKOVSKIY, P.A., prof., red.; KECHEK'YAN, S.V., prof., red.; POLYANSKIY, F.Ya., prof., red.; RYBNIKOV, K.A., prof., red.; SKAZKIN, S.D., akademik, red.; SOLOV'YEV, A.N., dotsent, red.; ZAITSEVA, M.G., red.; GEORGIYEVA, G.I., tekhn.red.

Petr Ivanovich Strakhov. Moskva, Izd-vo Mosk.univ., 1959.  
91 p. (MIRA 13:2)

(Strakhov, Petr Ivanovich, 1757-1813)

BARANOVSKIY, L.M., vrach; ZAYTSEVA, M.D., meditsinskaya sestra

Nurses' councils. Med. sestra 20 no.6:56-57 Je '61. (MIA 14:7)

1. Iz Gorodskoy klinicheskoy bol'nitsy No.2, Ryazan' (for Baranovskiy).
2. Iz Krovoozerskoy rayonnoy bol'nitsy Nikolayevskoy oblasti (for Zaytseva).

(RYAZAN--NURSES AND NURSING)  
(NIKOLAEVSK PROVINCE--NURSES AND NURSING)

GOVORUKHIN, A.P.; PSHENICHNAYA, A.M.; SMOLENSKAYA, T.V.; ZAITSEVA, M.B.;  
Prinimali uchastiye: KALASHNIKOV, N.V.; PLAKSINA, A.I.;  
DOLGOSHOV, V.I., starshiy nauchnyy otzudnik. PORTNOYAGIN, I.I.,  
otv.red.; ROGOVSKAYA, Ye.G., red.; BRAVININA, M.I., tekhn.red.

[Agroclimatic reference book on Orel Province] Agroklimaticheskii  
spravochnik po Orlovskoi oblasti. Leningrad, Gidrometeor.izd-vo,  
1960. 91 p.  
(MIRA 13:11)

1. Kurak. Gidrometeorologicheskaya observatoriya. 2. Upravleniye  
gidrometstuzhby tsentral'no-chernozemnykh oblastey (for Govorukhin,  
Pshenichnaya, Smelaya). 3. Institut geografii AN SSSR (for Dolgoshov).  
(Orel Province--Crops and climate)

GOVORUKHIN, A.P.; SMELEVA, T.V.; PSHENICHNAYA, A.M.; ZAITSEVA, M.B.  
Prinimali uchastiye: KALASHNIKOV, N.V.; PLAKSINA, A.I.;  
DOLGOSHOV, V.M., starshiy nauchnyy sotrudnik. PORTFAGIN,  
I.I., otv.red.; MIRONENKO, Z.I., red.; VOLKOV, N.V., tekhn.red.

[Agroclimatic manual for Lipetsk Province] Agroklimaticheskii  
spravochnik po Lipetskoi oblasti. Leningrad, Gidrometeor.izd-vo,  
1960. 94 p. (MIRA 14:1)

1. Russia (1923- U.S.S.R.) Glavnaya upravleniya gidrometeorologicheskoy sluzhby. Upravleniya gidrometeorologicheskoy sluzhby TSentral'no-Chernozemnykh oblastey. 2. Upravleniya gidrometsluzhby TSentral'no-Chernozemnykh oblastey (for Govorukhin, Smeleva, Pshenichnaya, Zaitseva). 3. Institut geografii Akademii nauk SSSR (for Dolgoshov).

(Lipetsk Province--Crops and climate)

GOROKHOV, D.I.; GOVORUKHIN, A.P.; SVELAYA, T.V.; PSHENICHNAYA, A.M.;  
ZAYTSEVA, M.B.; Prinimali uchastiye: KALASHNIKOV, N.V.;  
PLAKSINA, A.I., PORTNIAGIN, I.I., otv.red.; ROGOVSKAYA, Ye.G.,  
red.; VOLKOV, N.V., tekhn.red.

[Agroclimatic reference book on Tambov Province] Agroklimati-  
cheskii spravochnik po Tambovskoi oblasti. Leningrad, Gidro-  
meteor.izd-vo, 1959. 123 p. (MIRA 13:2)

1. Kurek. Gidrometeorologicheskaya observatoriya. 2. Upravle-  
niye gidrometstushby TSentral'no-Chernozemnykh oblastey (for  
Gorokhov, Govorukhin, Svelaya, Pshenichnaya, Zaytseva).  
(Tambov Province--Crops and climate)

GOVORUKHIN, A.P.; SHILAYA, T.V.; PSHENICHNAYA, A.M.; DOLGOSEV, V.I.,  
nauchnyy sotrudnik; ZAYTSEVA, M.B.; NEDOSHIVINA, T.G., red.;  
VLADIMIROV, O.G., tekhn.red.

[Agroclimatic manual for Bryansk Province] Agroklimaticheskii  
spravochnik po Bryanskoi oblasti. Leningrad, Gidrometeor.izd-vo,  
1960. 111 p. (MIRA 14:4)

1. Russia (1923- U.S.S.R.) Glavnoye upravleniye gidrometeorolog-  
cheskoy sluzhby. Upravleniye gidrometeorologicheskoy sluzhby  
TSentral'no-chernozemnykh oblastey. 2. Institut geografii AN SSSR  
(for Dolgoshev).

(Bryansk Province--Crops and climate)

KOZLOV, A.M.; ZAYTSEVA, M.A.; BOLDOV, V.G.

Our experience in compiling a reference book for the transcription  
of geographical names. Good. i kart. no.1145-48 in 4m.  
(EKA 170)

BELOV, K.P.; ZAYTSEVA, M.A.; KODOMTSHEVA, A.M.

Characteristics of magnetic hysteresis phenomena in the systems  
 $\text{Pr}_2\text{O}_3$ ,  $\text{Fe}_2\text{O}_3$ , and  $\text{La}_2\text{O}_3$ .  $\text{Fe}_2\text{O}_3$ . Zhur.eksp.i teor.fiz. 37  
no.4:1159-1161 O '59. (MIRA 13:5)

1. Moskovskiy gosudarstvenny universitet.  
(Praseodymium oxide--Magnetic properties)  
(Iron oxide--Magnetic properties)  
(Lanthanum oxide--Magnetic properties)

BELOV, K.P.; ZAYTSEVA, M.A.; KADOMTSEVA, A.M.

Magnetic properties of the lanthanum and praseodymium orthoferrites  
in the partial substitution of the  $Fe^{3+}$  ions for  $Al^{3+}$  ions. Zhur.  
eksp. i teor. fiz. 39 no.4; 1148-1150 O '60. (MIRA 13:11)

1. Moskovskiy gosudarstvennyy universitet.  
(Rare earth ferrates--Magnetic properties)

VYDREVICH, Ye. Z.; ZAYTSEVA, M.A.

Density of aluminate-alkali solutions. TSvet. met. 34  
no. 6: 55-58 Je '61. (MIRA 14:6)  
(Aluminum-Metallurgy)  
(Hydrometallurgy)

Magnetic anisotropy and . . .

30073  
S/046/61/025/011/010/031  
B117/B102

ASSOCIATION: Fizicheskiy fakul'tet Moskovskogo gos. universiteta im.  
M. V. Lomonosova (Physics Division of Moscow State  
University imeni M. V. Lomonosov)

Card 4/4

30073  
S/048/61/025/011/013/031  
B117/B102

Magnetic anisotropy and . . .

difference in the magnetization of the two sublattices of  $\text{Fe}^{3+}$  ions, when they are partly replaced by nonmagnetic  $\text{Al}^{3+}$  ions. However, the Curie point is lowered in every case. Compositions in which  $\text{Fe}^{3+}$  ions were partly replaced by magnetic  $\text{Cr}^{3+}$  ions were examined. As compared with stoichiometric compositions, the coercive force diminishes sharply in La and Pr orthoferrites, when  $\text{Cr}^{3+}$  ions are introduced. As expected, and unlike  $\text{Al}^{3+}$  ions,  $\text{Cr}^{3+}$  ions do not change magnetization very much. The Debye diagrams taken by A. A. Katsnel'son and K. Yatskul'yan showed that all the examined compositions are solid solutions without any foreign phase. Ye. A. Turov is thanked for his help and for having discussed the results obtained. V. A. Naysh (Fizika metallov i metallovedeniye, 2, 10 (1960); 11, 161 (1960)) is mentioned. There are 5 figures and 6 references: 2 Soviet and 4 non-Soviet. The three references to English-language publications read as follows: Bozorth, R. M., Phys. Rev. Letters, 1, 362 (1958); Gilleo, M. A., J. Chem. Phys. 24, 1239 (1956); Watanabe, H., J. Phys. Soc. Japan, 14, 511 (1959). X

Card 3/4

30073  
S/CIA/61/025/C11/015/11  
B117/B102

Magnetic anisotropy and . . .

orthoferrites. If the temperature is increased to the Curie point, the anisotropy of weak ferromagnetism is virtually not reduced. The characteristic phenomena of hysteresis and thermal remanence effects appearing in rare-earth orthoferrites can be explained also by the strong magnetic anisotropy. Thermal remanence phenomena were observed on polycrystalline La, Pr, and Yb orthosilicates (Ref. 5; Below V. P. Zaytseva, L. A. Matveeva, A. I., Zh. mineral., 1966, 10, 117 (1967); Ref. 6; Matsubae, H. J. Phys. Soc. Japan, 19, 911 (1964)). Magnetization curves and hysteresis loops were recorded in magnetic fields of up to 20,000 oe by a ponderomotive method. Specimens cooled in the magnetic field displayed asymmetric hysteresis loops with individual cycles. The thermal remanence was removed by a magnetic field of the order of 10,000 oe. This is indicative of the enormous coercive force of order of 10,000 oe. This is indicative of the enormous coercive force of these orthoferrites. A partial substitution of nonmagnetic Al<sup>3+</sup> ions for Fe<sup>3+</sup> ions was performed in polycrystalline La and Pr orthoferrites. A sharp diminution of both coercive force and thermal remanence phenomena was observed along with a steep rise of magnetization. The latter may be explained by a prevailing diminution of the exchange field. On the other hand, the growth of magnetization is possibly associated with the greater

Card 2/4

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S/048/61/025/011/013/031  
B117/B102

AUTHORS:

Belov, K. P., Zaytseva, M. A., Kadomtseva, A. M., and  
Timofeyeva, V. A.

TITLE:

Magnetic anisotropy and hysteresis properties of rare-earth orthoferrites

PERIODICAL: Akademiya nauk SSSR. Izvestiya Seriya fizicheskaya, No. 11,  
no. 11, 1961, 1389-1392

TEXT: Magnetic anisotropy was examined on single crystals of La, Pr, Nd,  
Sm, Eu, Gd, and Yb orthoferrites. The crystals were grown by spontaneous  
crystallization from their solution in a melt of lead compounds, lead  
oxide, and lead fluoride. The torque of the resulting crystals as a func-  
tion of their angle of rotation with respect to an external magnetic field  
of up to 20 koe was measured with an anisometer. The torque curves drawn  
at room temperature resembled one another in the examined single crystals  
and showed that the orientation of the magnetic moment in the axis of  
easiest magnetization is very stable against rotation of the outer field.  
This points to an exceedingly strong magnetic anisotropy of these

Card 1/4

X

S/070/62/007/002/008/022  
The magnetic properties and structures... E132/E160

that of the second 12.347 as compared with 12.387 for the pure Y Fe garnet. In garnet there are three magnetic sub-lattices and on Neel's model M the curve observed for the first composition can be satisfactorily explained if the lattice having a weak inherent exchange interaction takes a different course from that of the other (iron) sublattices. The Ti-containing garnets  $Mn_{0.5}Y_{2.5}Fe_{4.5}Ti_{0.5}O_{12}$  and  $MnY_2Fe_4TiO_{12}$  were examined but showed no anomalies except that the second compound had a "tail" of residual magnetisation which persisted above the Curie point ( $506^{\circ}\text{C}$ ) apparently connected with the appearance of another phase (traces of  $Y_2Ti_2O_7$  were observed in the X-ray powder photograph).

There are 4 figures.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im.  
M.V. Lomonosova  
(Moscow State University imeni M.V. Lomonosov)

SUBMITTED: June 27, 1961

Card 2/2

24.2200

AUTHORS: Belov, K.P., Zaytseva, M.A., Kadomtseva, A.M.,  
Kvitka, S.S., and Ovchinnikova, T.L.

TITLE: The magnetic properties and structures of certain  
garnet systems

PERIODICAL: Kristallografiya, v.7, no.2, 1962, 242-246

TEXT: Garnet structures have been synthesized by the  
substitution in yttrium iron garnets of Fe and Y ions by Mn, Ge  
and Ti and their structures and magnetic properties have been  
studied. In the garnet of composition  $Mn_{0.5}Y_{2.5}Fe_{4.5}Ge_{0.5}O_{12}$   
an anomalous temperature dependence of the spontaneous  
magnetisation has been observed at low temperatures (of Neel's  
type M). It is established that the garnet of composition  
 $Mn_{2}Fe_4GeO_{12}$  has a Curie point below 0 °C and that the curve of  
the temperature dependence of the spontaneous magnetisation tends  
asymptotically to zero. The curves are explained qualitatively.  
The cell size of the first-mentioned compound is 12.367 Å, and

Card 1/2

BELOV, K.P.; ZAYTSEVA, M.A.; KADOMTSEVA, A.M.; TIMOFEYeva, V.A.

Magnetic anisotropy and hysteresis properties of orthoferrates of rare earth elements. Izv. AN SSSR. Ser. fiz. 25 no.11:  
1389-1392 N '61. (MIRA 14:11)

1. Fizicheskiy fakul'tet Moskovskogo gosudarstvennogo universiteta im. M.V. Lomonossova.

(Rare earth ferrates--Magnetic properties)

BELOV, K.P.; ZAYTSEVA, M.A.; KADOMTSEVA, A.M.; KVITKA, S.S.; OVCHINNIKOVA,  
T.L.

Magnetic properties and structure of some garnet systems.  
Kristallografiia 7 no.2:242-246 Mr-Ap '62. (MIRA 15:4)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova.  
(Garnet--Magnetic properties) (Crystallography)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

VYDREVICH, Ye.Z.; ZAYTSEVA, M.A.; ORLOVA, A.N.

Diagrams for determining the density of aluminate sodium-alkali solutions. TSvet. met. 36 no.10:56-60 O '63.

(MIRA 16:12)

ACCESSION NR: AP4011744

is associated with change in degree of dispersion and with the orthorhombic distortion of the lattice. Along with these changes, an increase was observed in magnetization intensity. This is explained by the ordered distribution of Al<sup>+3</sup> ions in the crystal lattice. Orig. art. has: 3 figures.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova (Moscow State University)

SUBMITTED: 10Jul63 DATE ACQ: 14Feb64 ENCL: 00

SUB CODE: PH NO REF Sov: 002 OTHER: 002

Card 2/2

ACCESSION NR: AP4011744

S/0181/64/006/001/0101/0107

AUTHORS: Belov, K. P.; Iversonova, V. I.; Zaytseva, M. A.; Kadomtseva, A. M.; Katsnel'son, A. A.; Yatskull'yak, K.

TITLE: Magnetic and structural properties of lanthanum orthoferrite during partial replacement of Fe<sup>3+</sup> ions by other trivalent ions

SOURCE: Fizika tverdogo tela, v. 6, no. 1, 1964, 101-107

TOPIC TAGS: magnetic property, structural property, orthoferrite, lanthanum, lanthanum orthoferrite, Fe<sup>3+</sup>, Al<sup>3+</sup>, Sc<sup>3+</sup>, Co<sup>3+</sup>, thermoremanent magnetization, magnetization intensity, hysteresis loop, crystal lattice

ABSTRACT: In these studies the Fe<sup>3+</sup> ion was replaced, in part, by Al<sup>3+</sup>, Sc<sup>3+</sup>, Cr<sup>3+</sup>, and Co<sup>3+</sup>. Thermoremanent magnetization of LaFeO<sub>3</sub> cannot be reduced to zero even in a field of 20 000 oersteds, but if Al<sup>3+</sup> ions replace some of the Fe<sup>3+</sup> ions (LaFe<sub>0.9</sub>Al<sub>0.1</sub>O<sub>3</sub>), introduced by orthorhombic distortion of the crystal lattice, thermoremanent magnetization almost disappears, and the hysteresis loops become symmetrical. These changes may be explained by the finely dispersed character of the samples. The change in magnetic properties on substitution of the indicated ions

Card 1/2

BELOV, K.P.; IVERONOVA, V.I.; ZAYTSEVA, M.A.; KADOMTSEVA, A.M.; KATSNEL'SON, A.A.; YATSKUL'YAK, K.

Magnetic and structural properties of lanthanum orthoferrite  
with partial substitution of  $Fe^{+3}$  ions by other trivalent ions.  
Fiz. tver. tela 6 no.1:101-107 Ja '64. (MIRA 17:2)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova.

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

2  
1000-10000 m.s. The first 1000 m.s. of the plateau is flat, leading to the  
second 1000 m.s. which is covered with small shrubs. The temper-  
ature is about 10°C. The plateau is situated in the region of the  
Tigris River, which flows through the plateau. The plateau is  
surrounded by mountains, and the plateau itself is very high.

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19. The following table shows the number of hours worked by 1000 workers in a certain industry.

10. *Leucosia* *leucostoma* *leucostoma* *leucostoma* *leucostoma*

100

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

On the Magnetic Properties of Oxygen Compounds of  
Gadolinium

SOV/56-36-6-7/66

ASSOCIATION: Moskovskiy gosudarstvennyy universitet ( Moscow State Uni-  
versity)

SUBMITTED: January 5, 1959

Card 4/4

On the Magnetic Properties of Oxygen Compounds of  
Gadolinium

Sov/56-36-6-7/66

anomaly in the behavior of the temperature dependence of magnetostriiction. Further results obtained by investigations concern gadolinium ferrite-perovskite  $Gd_2O_3 \cdot Fe_2O_3$ . Figure 8 shows the dependence of magnetization on the field (up to  $H = 7000$  Oe) for various temperatures between 18 and  $598^{\circ}C$ , and figure 9 shows the analogous magnetization isothermal lines, but after heating beyond Curie point in the magnetic field. Figure 10 shows the temperature dependence of spontaneous magnetization in the magnetic field after the first and second heating (the curves differ considerably). It is found that perovskite gadolinium ferrite possesses a weak ferromagnetism of the hematite type. Finally, the results obtained by an investigation of gadolinium-manganite (perovskite) are described. Figure 11 shows the  $H$ -dependence of magnetization at various temperatures, and figure 12 the hysteresis in  $Gd_2O_3 \cdot Mn_2O_3$  at  $4.3^{\circ}K$ , which may be observed within this temperature range although gadolinium manganite otherwise has paramagnetic properties. There are 12 figures, 1 table, and 6 references, 4 of which are Soviet.

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On the Magnetic Properties of Oxygen Compounds of  
Gadolinium

SOV/56-56-7/66

magnetization at helium temperatures and at Curie point ( $\theta = 561^\circ\text{K}$ ), coercive force, magnetostriction, etc. The results obtained by the investigations are shown by numerous diagrams and are discussed in detail. Figure 1 shows the temperature dependence of specific magnetization at various field strengths ( $H = 25.8, 129$  and  $1550$  Oe), figure 2 shows the temperature dependence of  $\sigma_s/\sigma_0$  and of the residual magnetization  $\sigma_r/\sigma_0$  within the range of compensation point, figure 3 shows the temperature dependence of the coercive force, and figure 4 the temperature dependence of the susceptibility of the paraprocess in  $3\text{Gd}_2\text{O}_3 \cdot 5\text{Fe}_2\text{O}_3$ ; figure 5 shows the temperature dependence of magnetostriction, figure 6 the dependence of  $(\sigma_s/\sigma_0)^2$  on  $(T/\theta)$  within the range of the Curie point (straight line), and figure 7 the dependence of the magnetization on  $H^{1/3}$  within the range of the Curie point. In a table the data of the garnet investigated are compared with those of other ferrimagnetics. It is found that at the compensation point and Curie point there is an anomalous growth of the coercive force and a very small paraprocess in garnet-ferrite and also an

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24(3)

SOV/56-36-6-7/66

AUTHORS: Belov, K. P., Zaytseva, M. A., Ped'ko, A. V.

TITLE: On the Magnetic Properties of Oxygen Compounds of Gadolinium  
(O magnitnykh svoystvakh okisnykh soyedineniy gadoliniya)

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1959,  
Vol 36, Nr 6, pp 1672 - 1679 (USSR)

ABSTRACT: Considerable interest is at present being displayed in the magnetic properties of the oxides (ferrites) of rare earths. The authors of the present paper investigated the temperature dependence of the magnetic properties of various gadolinium oxides; the samples were of garnet- or perovskite structure and were, contrary to what was the case in earlier investigations (Refs 1,2) sufficiently large, so that the data obtained were more accurate. The samples were tempered in air at 900°C for 6 hours, pressed into shape (block 60.5.5 mm) under high pressure, after which they were again tempered for 4 hours at 1300°C. The magnetic properties were measured by ballistic, magnetometric and ponderomotoric means. Ganolinium ferrite garnets were subjected to the closest investigation. The authors operated with  $3\text{Gd}_2\text{O}_3 \cdot 4.8\text{Fe}_2\text{O}_3 \cdot 0.2\text{Y}_2\text{O}_3$ . They investigated saturation

Card 1/4

The Magnetic- and Resonance Properties of the  
Ferrite Granates of Yttrium in the Substitution of  
 $Fe^{3+}$ -Ions by  $Cr^{3+}$ - and  $Al^{3+}$ -Ions

507/56-56-66/76

The ratios found agree qualitatively with the theory developed by Clogston et al. (Ref 4), i. e. that  $\Delta H$  is proportional to  $V_{\theta}^{\omega}$  and  $\theta$ . There are 2 figures and 4 references.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet (Moscow State University)

SUBMITTED: February 12, 1959

Card 3/3

The Magnetic- and Resonance Properties of the  
Ferrite Granates of Yttrium in the Substitution of  
 $Fe^{3+}$ -Ions by  $Cr^{3+}$  - and  $Al^{3+}$ -Ions

SOV/56-36-5-66/76

resonance-characteristics were carried out on crystalline samples (sintering in air at  $1300^{\circ}C$  for 4 hours, density  $2.75 \text{ g/cm}^3$ ). Figure 1 shows the connection between  $a$  and the saturation magnetization  $\sigma_o$  as well as between  $a$  and Curie point  $\theta$  in the range  $0 \leq a \leq 1$ . All four curves ( $\sigma_o$ ,  $\theta$  for  $Al^{3+}$  and  $Cr^{3+}$ ) show a more or less steep decline with increasing  $a$ , with the exception of the chromium-substituted sample which shows an incline at  $a < 0.5$  for  $\sigma_o$ . Figure 2 shows the results obtained by measurements of the width of the absorption lines  $\Delta H$ . With increasing  $a$  there is an increase of  $\Delta H$  for the chromium-substituted sample, and a decrease for the  $Al$ -substituted sample. For the former the  $g$ -factor increases from  $2.150 \pm 0.005$  (unsubstituted sample) to  $2.200 \pm 0.005$ , in the case of the latter it increases to  $2.030 \pm 0.005$ .

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24(2), 24(3)

AUTHORS: Belov, K. P., Zaytseva, M. A.,  
Malevskaya, L. A. SOV/58-36-3-66/76

TITLE: The Magnetic- and Resonance Properties of the  
Ferrite Garnets of Yttrium in the Substitution of  
 $Fe^{3+}$ -Ions by  $Cr^{3+}$ - and  $Al^{3+}$ -Ions (Magnitnyye i  
rezonansnyye svoystva ferritov-granatov ittriya pri  
zameshchenii ionov  $Fe^{3+}$  ionami  $Cr^{3+}$  i  $Al^{3+}$ )

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1959,  
Vol 36, Nr 5, pp 1602-1603 (USSR)

ABSTRACT: The present "Letter to the Editor" is in close connection  
with the preceding one (pp 1600-1601). The present letter  
deals mainly with the investigation of the influence  
exerted by foreign ions on the physical character. The  
stoichiometric compound  $3Y_2O_3 \cdot 5Fe_2O_3$  is conveyed to  
 $3Y_2O_3 \cdot (5-a)Fe_2O_3 \cdot aAl_2O_3$  and  $3Y_2O_3 \cdot (5-a)Fe_2O_3 \cdot aCr_2O_3$   
respectively by the substitutions. a denotes the content  
of  $Al^{3+}$  and  $Cr^{3+}$  ions. Measurements of the magnetic.. and

Card 1/3

AUTHORS: Belov, K. P., Zaytseva, M. A. SOV/53-66-1-6/11

TITLE: New Magnetic Materials - Ferrite-Garnets (Novyye magnitnyye materialy - ferrity-granaty)

PERIODICAL: Uspekhi fizicheskikh nauk, 1958, Vol. 66, Nr 1,  
pp. 141 - 144 (USSR)

ABSTRACT: The authors in extracts give the contents of a number of foreign papers (mainly from the USA and from France) dealing with ferrites of the formula  $3 \text{Me}_2\text{O}_3 \cdot 5 \text{Fe}_2\text{O}_3$  (or  $\text{Me}_3\text{Fe}_2\text{Fe}_3\text{O}_{12}$ ). There are 4 figures, 1 table, and 11 references.

1. Magnetic materials
2. Ferrites--Magnetic properties
3. Garnets--Magnetic properties

On Magnetic Properties of Ferrites Exhibiting  
a Compensation Point

SOV/48-22-10-23/23

for valuable suggestions. There are 10 figures, 3 tables,  
and 14 references, 4 of which are Soviet.

ASSOCIATION: Fizicheskiy fakul'tet Moskovskogo gos.universiteta imeni  
M. V. Lomonosova (Dept. of Physics at the Moscow State  
University imeni M. V. Lomonosov)

Card 4/4

USCOMM-DC-60,966

On Magnetic Properties of Ferrites Exhibiting  
a Compensation Point

SOV/48-22-10-23/23

interaction within the sublattice B compared with the interaction between the sublattices A and B must not be neglected. The measurements showed that the value of the absolute saturation in the system LiFeCr-ferrites becomes higher in the case of hardening. In technical publications there are data on the influence of hardening upon  $\sigma$  of various simple and composed ferrites (Refs 10 and 11)<sup>o</sup> and theories (Refs 12 - 14) explaining the results of the papers (Refs 10 and 11). According to this  $\sigma$  depends on the distribution of the cations on A and B. This distribution, however, depends on the temperature. In the present case the problem became more complicated as apart from the cation distribution also the variation of the angles between the magnetic moments in sublattices was possible. The possible influence of these two factors excludes a comparison of the experimental values found of saturation in hardening with respective theories. The question of the influence of these factors probably might be answered by means of radiographic and especially neutronographic investigations. The authors express their gratitude to K. G. Khomyakov and T. I. Bulgakova

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On Magnetic Properties of Ferrites Exhibiting  
a Compensation Point

SOV/48-22-10-23/23

the limiting cycle in the temperature range of from -30° to about 10 to 20° above the compensation point (by the astatic magnetometer); 4) temperature dependence of the paramagnetic sensitivity (according to the ponderomotive method). The measuring results showed that the ferromagnetic spinels Li<sub>x</sub>Fe<sub>1-x</sub>Cr<sub>y</sub> in a certain range of solution exhibit an anomalous shape of the curve σ(T) with a compensation point. This has been predicted by Neel. In contrast to the theory it was found that the compensation never was perfect. The phenomenon of an imperfect compensation may be explained by the heterogeneity of the samples. An other considerably greater difference is that the value of the absolute saturation computed (according to Neel) from the distribution of the cations does not agree at all with data found experimentally (Table 2, column 3 and 5). The modification of Neel's theory suggested by Yafet and Kittel (Ref 8) is capable of explaining this discrepancy qualitatively. The explanation is as follows: As the measured value of the magnetic value in these ferrites is lower than the value computed according to Neel's theory and M<sub>B</sub> > M<sub>A</sub>, in this case the negative exchange

Card 2/4

AUTHORS: Belov, K. P., Bol'shova, K. M., SOV/48-22-10-23  
Yelkina, T. A., Zaytseva, M. A.

TITLE: On Magnetic Properties of Ferrites Exhibiting a Compensation Point (O magnitnykh svoystvakh ferritov s tochkoy kompensatsii)

PERIODICAL: Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, 1958,  
Vol 22, Nr 10, pp 1282 - 1292 (USSR)

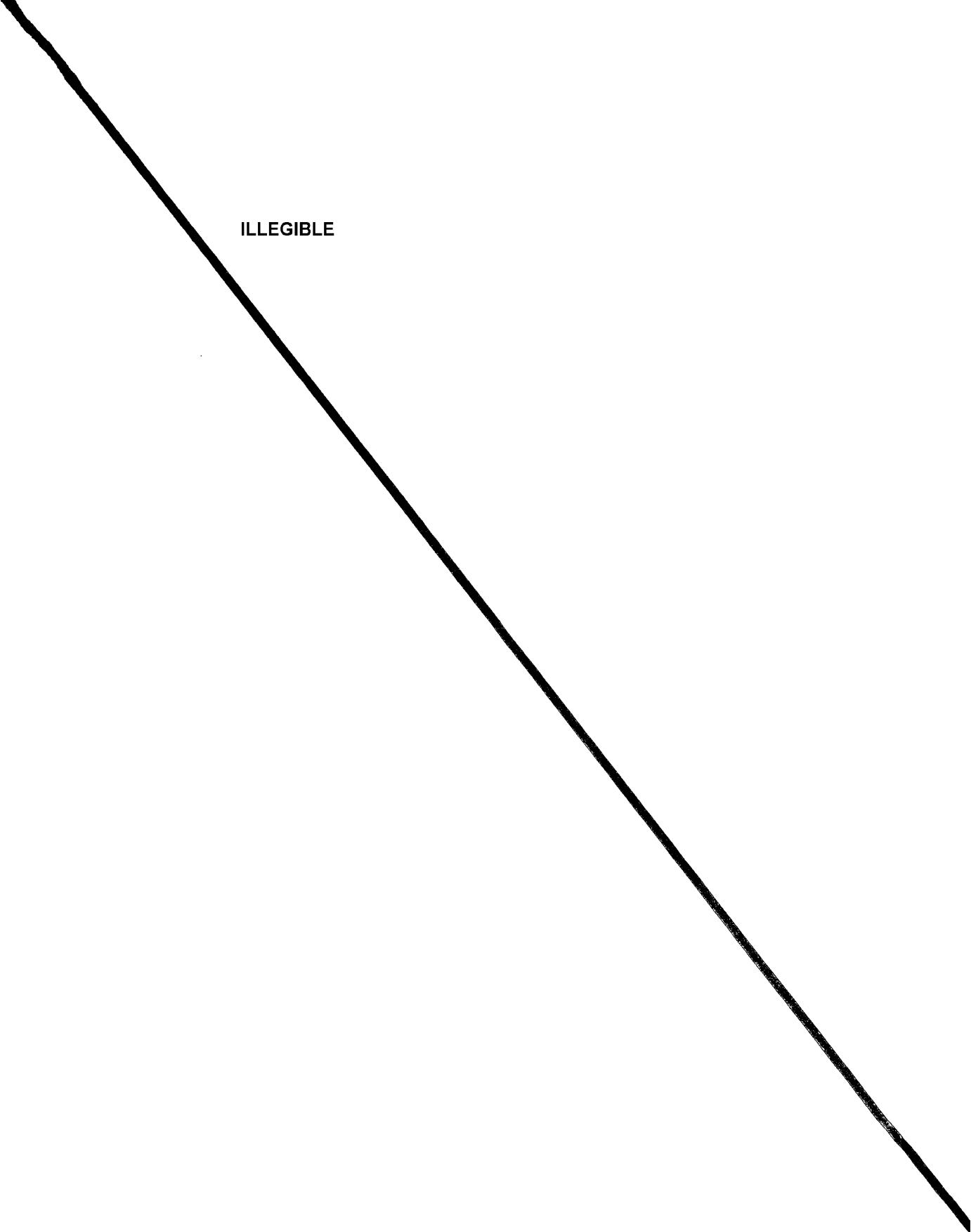
ABSTRACT: In the present paper the authors performed exact measurements of the magnetic properties of mixed lithium chromite ferrites (which were annealed and hardened) in the case of different annealing after hardening. For the investigation a system of ferromagnetic lithium spinels that contained chromium of the common formula  
$$\text{Li}_2\text{O} \cdot (5 - 2a) \text{Fe}_2\text{O}_3 \cdot 2a \text{Cr}_2\text{O}_3$$
 (for  $a = 1,25; 1,5; 1,6; 1,7$ ) was synthesized. The following magnetic characteristics were investigated: 1) Temperature dependence of the spontaneous magnetization of  $\sigma_0$  (T); 2) magnetic moments of the atoms (the measurements were carried out by A. V. Fed'ko); 3) temperature dependence of the residual magnetization of

*ZAYTSEVA, M.A.*

- 24(3)      Author: Drakow, G.P., Candidate of Physical-Mathematical Sciences      ZTF/54-59-2-14/55
- TITLE: Survey of Papers Read by Participants of Congress on Physics of Magnetic Materials in the All-Union Conference on Physics of Magnetic Materials (Obozr dokladov uchebno-issledovatel'skogo vystoyschashchego soveshchaniia po fizike magnitnykh materialov)
- PERIODICAL: Vestnik Moskovskogo Universiteta. Matematika, mehanika, astronoma, fizika, khimiya, 1959, № 1, pp. 1-22, 237-250 (1958)
- ABSTRACT: From December 6-11, 1957 there took place the Fourth Union Congress on Physics of Magnetic Materials in Moscow. (The first two meetings 1956 and 1957 in Leningrad, the third meeting 1956 in Krasnoyarsk). The conference was organized by the Academy of Sciences of the USSR, the Central Committee of Mathematical Sciences, Scientific Council of the Institute of Problems of Magnetism, Institute for Strength Problems of the Academy of Sciences, USSR and Committee for Materials Science. There were more than 100 participants from 14 countries. There among the following lectures were held: 1. On the properties of the magnetic materials given by: 1. Professor B.P. Telezhin, re. v. Kandidat Fiz.-Mat. Nauk - Magnetic Properties of the Ferromagnetic Elements; 2. Professor B.P. Telezhin, re. v. Kandidat Fiz.-Mat. Nauk - On Magnetic Viscosity of Ferromagnetic Materials; 3. Professor I.V. Kostrikov, re. v. Kandidat Fiz.-Mat. Nauk - Effect of Magnetic Viscosity on the Propagation Characteristics of Waves;
4. M.Y. Dzhigar, Lecturer "Variations of Structure and Anisotropic Magnetic Properties of NiFe";
5. M.M. Granovsky, Lecturer, Dept. Prochnost, Junior Scientific Assistant - Magnetic Properties of Antiferromagnetic Oxides;
6. G.P. Drakov, Lecturer "Magnetization Properties of Binary Alloys";
7. Professor Ya.I. Temerovsky, L.T. Slobodov, Assistant "Electric Properties of Metal Perites";
8. M.Z. Chikarov, Senior Scientific Assistant, A.P. Parshov, Assistant - Magnetic Properties and Structure of Manganese-Boron - Alloys;
9. N.M. Sogolov, Senior Scientific Assistant, B.P. Belov "Some Properties of Perites";
10. S.A. Golikov, Senior Scientific Assistant, Yu.N. Gerasimov, Lecturer "Properties of Ni-Pt, Cr-Ni, Cr-Cu, Cr-Al, Cr-Al<sub>2</sub>O<sub>3</sub>, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni, Cr-Al<sub>2</sub>O<sub>3</sub>-Cu, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al, Cr-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Cr-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
11. E.A. Danilov and Yu.S. Ponomarenko "Properties of Perites in the High Temperature Range";
12. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
13. E.P. Belov, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
14. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
15. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
16. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
17. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
18. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
19. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
20. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
21. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
22. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
23. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
24. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
25. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
26. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
27. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
28. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
29. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
30. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
31. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
32. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
33. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
34. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Ni);
35. Professor I.P. Slobodov, Senior Scientific Assistant, T.S. Tolstina, Lecturer "Properties of the Magnetic Materials Based on the Alloys Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al<sub>2</sub>O<sub>3</sub>, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>O<sub>3</sub>-Cu-Al, Ni-Al<sub>2</sub>O<sub>3</sub>-Ni-Al<sub>2</sub>

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

ILLEGIBLE



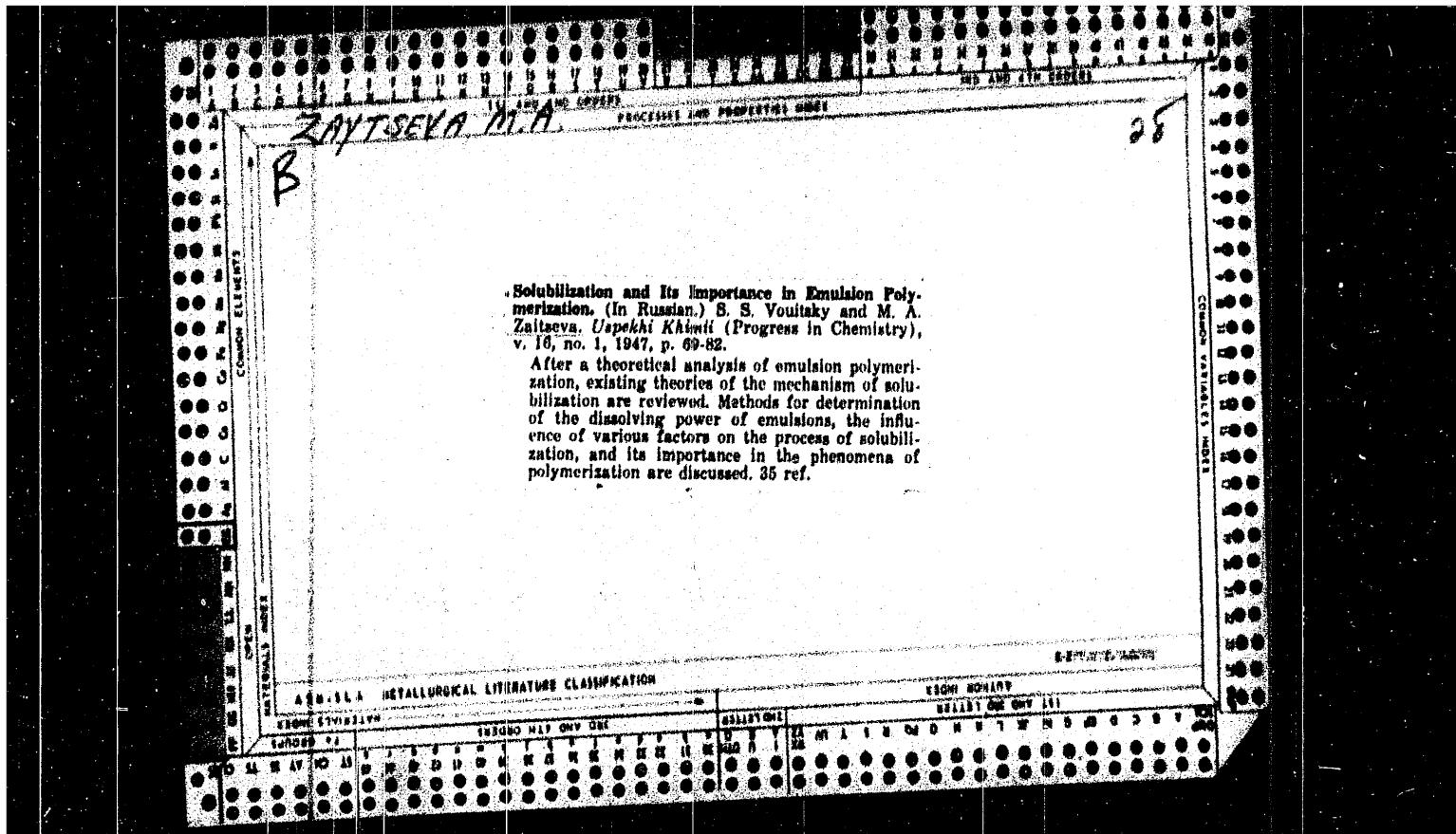
APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

Dissertation: "Influence of the Nature and Structure of Adsorbents on Adsorption from Solutions of Aliphatic Alcohols, Acids and Soaps."

28 Feb 49

Moscow Inst of Fine Chemical Technology imeni M. V. Lomonosov.

SO Vecheryaya Moskva  
Sum 71



ZAYTSEVA, M.

In the Presidium of the Ukrainian Academy of Sciences. Visnyk  
AN URSR 24 D '53. (MLRA 7:3)  
(Ukraine--Science) (Science--Ukraine)

GOLOVINA, Lidiya Ivanovna; YAGLOM, Isaak Moiseevich; ZATSEVA, M., redaktor;  
AKHLMOV, S.N., tekhnicheskij redaktor

[Induction in geometry] Induktsiya v geometrii. Moskva, Gos. izd-vo  
tekhniko-teoret. lit-ry, 1956. 98 p. (Populiarnye lektsii po matema-  
tike, no.21) (MLRA 9:11)  
(Geometry) (Induction (Mathematics))

ZAYTSEVA, M.

1. ZATTSEVA, M.
2. USSR (600)
4. Academy of Sciences of the Ukrainian S. S. R.
7. In the Presidium of the Academy of Sciences of the Ukrainian S. S. R.  
Visnyk AN URSR, 24, no. 1, 1953.

9. Monthly List of Russian Accessions, Library of Congress, May 1953, Unclassified.

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

ZAITSEVA, M.

In the Presidium of the Academy of Sciences of the U.R.S.R. Visnyk AN UkrSSR  
24 no.9:77-79 S '53.  
(MLRA 6:10)  
(Academy of Sciences of the Ukrainian SSR)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

XAYTSYVA, M.

In the Presidium of the Academy of Sciences of the Ukrainian S.S.R.  
Visnyk AN UkrSSR 24 no.11:76-77 N 152.  
(MIRA 6:12)  
(Academy of Sciences of the Ukrainian S.S.R.)

USSR/Cultivated Plants - Fruits. Berries.

M

Abs Jour : Ref Zhur Biol., No 18, 1958, 32529

Author : Kuminov, Ye. Zaytseva, M.

Inst : -

Title : Wild Growing Berry Fields of Tuva

Orig Pub : S. kh. Sibiri, 1957, № 12, 95-98

Abstract : Minusinsk, Fruits and Berry Experiment Field revealed wild growing mountain ash, hawthorn, elder, dog rose, the raspberry (*Rubus saxatilis*), blueberry, bilberry, cranberry, cultivated strawberry, the wild strawberry (*Fragaria vesca*), gooseberry of different forms resistant to *Sphaerotheca*, black currant with large berries, red currant, raspberry (*Rubus idaeus*) and the common seabuckthorn. A number of forms of wild growing berry fields is of interest for introduction in cultivation.

Card 1/1

SAYFULLIN, R.S., ZAYTSEVA, L.V.

Electrophoretic separation of inorganic substances from organic media. Zhur. fiz. khim. 39 no. 7/13/20/4(23) J1 '65.  
(MIRA 18:8)

1. Kazanskly khimiko-tehnologicheskiy institut.

SAYFULLIN, R.S.; ZAITSEVA, L.V.

Electrophoretic deposition of capron. Kell. zhur. 27 no.5:755-757  
S-0 '65. (MIRA 18:10)

1. Kazanskiy khimiko-tehnologicheskiy institut imeni Kirova.

L 1812-66

ACCESSION NR: AP5024022

Such coatings will protect metals from corrosion and mechanical action. Orig.  
art. has: 4 figures.

5

[SM]

ASSOCIATION: Kazanskiy khimiko-tehnologicheskiy institut im. S. M. Kirova (Kazan  
Chemical Technology Institute)

SUBMITTED: 19Jun64

ENCL: 00

SUB CODE: MT, EM

NO REF Sov: 005

OTHER: 007

ATD PRESS: 4/11

Card 2/2

L 1812-66 EWT(m)/EFF(c)/EWP(j)/T DJ/RM  
ACCESSION NR: AP5024022

UR/0069/65/027/005/0755/0757  
541.18.047.6

AUTHOR: Sayfullin, R. S.; Zaytseva, L. V.

TITLE: Electrophoretic deposition of capron

SOURCE: Kolloidnyy zhurnal, v. 27, no. 5, 1965, 755-757

TOPIC TAGS: plastic coating, protective coating, nylon/capron

ABSTRACT: A study has been made of the electrophoretic deposition of capron (polycaprolactum) to form protective coatings on metals.<sup>15</sup> This work was done because electrophoretic deposition eliminates the disadvantages of conventional methods of coating with poorly soluble polymers. The electrophoretic behavior of capron powder (particle size, 2-10 micron) in organic and aqueous media was studied and the optimum process conditions and suspension compositions were determined. It was found that deposition usually occurs on the cathode, but sometimes on both electrodes. Organic media were found to be most suitable for the deposition, toluene being the best. Strongly adhering coatings 100-200 micron in thickness were produced on Fe, Cu, Al, Ni, Zn, and brass in 3 min by electrophoresis from toluene with subsequent heat treatment at 220-240C for 3-10 sec.

Card 1/2

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

doktor tekhn. nauk, retsenzent; SMIKNUVA, A.V., kand. tekhn. nauk, retsenzent; FOMIN, N.V., red.; GORDON, L.M., red. izd-va; ISLENT'YEVA, P.G., -tekhn. red.

[Electrolytic polishing and pickling of metallographic sections] Elektropolirovanie i elektrotravlenie metallograficheskikh shlifov. 2., perer. izd. Moskva, Metallurgizdat, 1963. 410 p. (MIRA 16:5)

(Metallography--Equipment and supplies)

(Electrolytic polishing)

(Metals--Pickling)

**Electrolytic polishing of metallurgical sections.** I. P. Stepanov and L. Ya. Popkov. *Zavodskaya Lab.*, 13, 679-92 (1946) (in Russian). - With c.d. 10-100 amp./sq. dm., 1-60 min., temp. 15-25 and 70-90°, neither solns. of  $H_3PO_4$  (50-93%), with or without addn. of citric acid, glycerol or oilale, nor solns. of  $H_2SO_4$  (24-68%), with or without addn. of glucose, peptone, or citric acid, proved unsatisfactory; pitting occurred in all samples. Satisfactory results were obtained with mixts. of  $H_2SO_4$  (12-10%) and  $H_3PO_4$  (51-98%) with addns. of glycerol, citric acid, lactic acid, or casein.  $H_3PO_4$  81,  $H_2SO_4$  12,  $H_2O$  6.5, glycerol 24, or  $H_3PO_4$  97.9,  $H_2SO_4$  10.3,  $H_2O$  10.7, casein 3.1, etch stainless, acid-resistant and high-speed steel at room temp., and polish when hot. Mixts. of  $H_3PO_4$  (70-81.5%),  $H_2SO_4$  (11.5-17%), and  $CrO_3$  (3.5-6.5%), without addn., act essentially in the same way, on all steels, both C and alloy; these electrolytes are not, however, effective when freshly prep'd, but improve on working up to a certain stage, after which their efficiency and etching power fall off rapidly. The condition of the  $H_3PO_4$ - $H_2SO_4$ - $CrO_3$  bath can be diagnosed by visual colorimetric comparison with a standard  $Cr^{III}$ - $Cr^{VI}$  scale: the bath polishes in the medium interval  $Cr^{III}/Cr^{VI} = 1:1$  to 4:1, etches at 1:4 and 9:1 and corrodes at more extreme values of that ratio; these limits are valid for a soln. of sp. gr. 1.6-1.7;  $\eta$ , 7-20 centipoises. Artificial conditioning of the freshly prep'd.  $H_3PO_4$  76.5,  $H_2SO_4$  11.5,  $H_2O$  0,  $CrO_3$  0 electrolyte was attempted by passing a current in the cold and hot soln. and by evapg. at 120°; the latter procedure proved most effective. A bath requiring but a very short preliminary conditioning can be prep'd. directly from the anhydrous ingredients:  $H_2SO_4$  (d. 1.84) 12.1,  $H_3PO_4$  (d. 1.54) 81.3,  $CrO_3$  (solid) 0.6%. A bath having departed from the correct sp. gr.,  $\eta$  and  $Cr^{III}/Cr^{VI}$  criteria in the course of operation, is best re-

generated by electrolytic oxidation with the Pb tank as anode and an immense porous vessel containg a Pb cathode, at an anodic c.d. 0.5 amp./sq. dm., until the correct characteristics are reached; the etching-polishing ability of the bath is then fully restored. Regeneration by addn. of  $CrO_3$ , upsetting the total Cr content, or by chem. oxidation with  $(NH_4)_2S_2O_8$ , is not recommended. The correct  $H_3PO_4$ - $H_2SO_4$ - $CrO_3$  of the given compn. is universal; best conditions for polishing are, for C steels, 30-40 amp./sq. dm., for medium-alloy steels 10-60, for high-alloy 60-70; temp., 75-90°, 8-15 min. The medium  $Cr^{III}/Cr^{VI}$ , normally polishing electrolyte, can be made to etch by lowering the c.d. to 15-30 amp./sq. dm. and the temp. to 40-60°. The c.d., time, and temp. conditions are const. for a given steel, irrespective of its heat-treatment, except for a slight increase of pitting in annealed and highly tempered samples. After unsatisfactory etching, samples can be repolished and re-etched. Drawbacks of the electrolytic methods are curvature of the polished surface, leaching out of certain inclusions and components, some amt. of opaque or colored film, striation and spotting. Main advantages are absence of mechanical and thermal changes and high fineness of structural detail.

W. R. Henn

MASLENNIKOVA, N.L.; YAKUBOVICH, S.V.; SANZHAROVSKIY, A.T.; RIVLINA, Yu.L.;  
Prinimali uchastiye: EMMANUILOV, Yu.M.; KRUCHININA, G.I.;  
ZAYTSEVA, L.V.

Internal stresses developed in the process of formation  
and aging of nitrocellulose coatings. Lakokras.mat.i ikh prim.  
no.1:15-18 '63. (MIRA 16:2)

(Paint materials)  
(Strains and stresses)

YAKUBOVICH, S.V.; MASLENNIKOVA, N.L.; Prinimalli uchastiye: ZAYTSEVA,  
L.V.; KRUCHININA, G.I.

Investigating the adhesion of paint coatings under aging con-  
ditions. Lakokras.mat. i ikh prim. no.4:20-25 '62. (MIRA 16:11)

ZAYTSEVA, L.V., starshaya meditsinskaya sestra

Treatment of tuberculosis of the lungs and respiratory tracts with aerosol inhalations. Med. sestra 20 no.9:44-47 8 '61.

(MIMA 14:10)

1. Iz Moskovskogo nauchno-issledovatel'skogo instituta tuberkuleza  
Ministerstva zdravookhraneniya RSFSR.

(AEROSOL THERAPY) (INHALATION (THERAPY))  
(TUBERCULOSIS)

ZAYTSEVA, L.V.; RAFALIBEYLI, R.Kh.

Structural characteristics of the Mekhmane ore zone and the stages  
of ore formation. Zakonom.razm.polezn.iskop. 7:363-364 '64.

(MIRA 17:6)

1. Upravleniye geologii i okhrany nedr pri Sovete Ministrov  
AzerbSSR.

KVYATKEVICH, I.K., kand.tekhn.nauk, dotsent; ARBUZOV, S.V., kand.tekhn.nauk;  
Prinimali uchastiye: KRASIKOVA, Z.N.; NASYROVA, Sh.I.;  
SOLOV'YEV, N.S.; SHILOVA, Z.F.; ZAYTSEVA, L.V.; KOROTKOVA, L.N.;  
KONYILKIN, A.F.; GLAMAZDA, V.P.; LOZHKOVA, V.T.

New simplified method of leather drying and moisturizing.  
Izv.vys.ucheb.zav.; tekhn.leg.prom. 3:43-58 '62. (MIRA 15:6)

1. Vsesoyuznyy zaochnyy institut tekstil'noy i legkoy  
promyshlennosti (for Kvyatkevich). 2. Tsentral'nyy nauchno-  
issledovatel'skiy institut kozhevenno-obuvnoy promyshlennosti  
(for Arbuzov). Rekomendovana kafedroy mashin i avtomatov  
Vsesoyuznogo zaochnogo instituta tekstil'noy i legkoy promysh-  
lenosti.

(Leather---Drying)

ZAYTSEVA L.S.

Thermodynamic calculation of some reduction reactions of barium  
oxide. Izv. AN SSSR. Ser. fiz. 20 no.10:1123-1126 O '56.

(MLRA 10:1)

1. Institut fiziki Akademii nauk USSR.  
(Barium oxides) (Electron tubes)

ZAYTSEVA, L.S.

123 - 1 - 251 D

Translation from: Referativnyy Zhurnal, Mashinostroyeniye, 1957,  
Nr 1, p. 43 (USSR)

AUTHOR: Zaytseva, L.S.

TITLE: Experimental Study on the Heat Conductivity of Mono-  
atomic Gases and Aviation Fuel Vapors (Eksperimental'-  
noye issledovaniye teploprovodnosti odnoatomnykh  
gazov i parov aviatsionnykh topliv)

ABSTRACT: Bibliographic entry on the author's dissertation for  
the degree of Candidate of Technical Sciences, pre-  
sented to the Moscow Aeronautical Institute (Mosk.  
aviats. in-t), Moscow, 1956

ASSOCIATION: Moscow Aviation Institute (Mosk. aviats. in-t)

Card 1/1

ZAYTSEVA, L. S.

The growth rate and V/I curves of  $\text{Co}_x\text{O}$  and  $\text{Co}_x\text{O}_y$  mixtures of  $\text{Co}_x\text{O}$  were measured. The results of the measurements on the coated arcs show that the concentration of  $\text{PO}_4^{3-}$  of alkali salts of carbonates in vapors ( $\text{Li}_2\text{CO}_3$ ,  $\text{Na}_2\text{CO}_3$ ,  $\text{K}_2\text{CO}_3$ ,  $\text{CaCO}_3$ ,  $\text{MgCO}_3$ ) has no influence on the rate of formation of  $\text{Co}_x\text{O}_y$ . The rate of formation of  $\text{Co}_x\text{O}_y$  depends upon the ratio between the  $\text{C}$  particles formed and the  $\text{O}_2$  ions, which are  $\text{M}_2\text{O}^{+}$  (where  $\text{M}$  is the metal ion, resp.). The presence of trace elements in the arc prevents the formation of secondary

VARGAFTIK, N.B.; TAYTSEVA, L.S.

Heat conductivity of D<sub>2</sub>O in the gaseous phase. Inzh.-fiz. zhur.  
6 no.5:3-6 My '63. (MIRA 16:5)

1. Aviatsionnyy institut imeni Sergo Ordzhonikidze, Moskva.  
(Deuterium oxide--Thermal properties)

ACCESSION NR: AP3040437

ENCLOSURE: 3

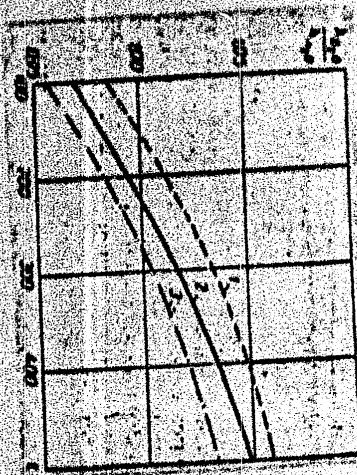


Fig. 3 Ratio  $\lambda_{D_2O} / \lambda_{H_2O}$  as a function of temperature  $t$  ( $^{\circ}\text{C}$ ) according to experimental and calculated data. 1,3 - calculated, 2 - experimental.

Card 5/5

ACCESSION NR.: AF3000-37

ENCLOSURE: 2

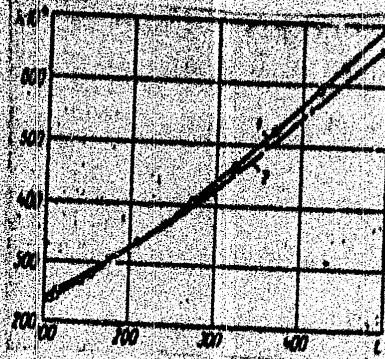


Fig. 2 Heat conductivity  $\lambda$  (J/m·sec·degree) of  $D_2O$  and  $H_2O$  vapor as a function of the temperature  $t$  ( $^{\circ}C$ ) according to experimental data: 1 -  $D_2O$ , 2 -  $H_2O$ .

Card 4/5

ACCESION NR: AIP3001437

ENCLOSURE: 1

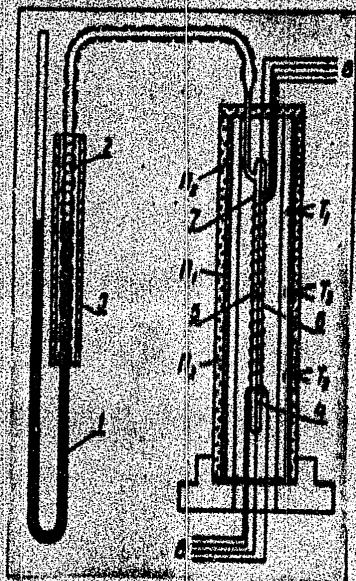


Fig. 1 Experimental apparatus.

1 - mercury, 2 - liquid investigated,  
3 - electric heater, 4 - tungsten spring,  
5 - platinum heater, 6 - external resist-  
ance thermometer, 7 - measuring tube,  
8 - leads to potentiometer, H<sub>1</sub>, H<sub>2</sub>, H<sub>3</sub> -  
thermostat heaters, T<sub>1</sub>, T<sub>2</sub>, T<sub>3</sub> - thermo-  
couples.

Card 3/5

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

ACCESSION NR: AP3000437

ASSOCIATION: Aviatcionnyy institut im. Sergo Ordzhonikidze (Aviation Institute im. Sergo Ordzhonikidze), Moscow

SUBMITTED: 24Sept62 DATE ACQ: 10Jun63 ENCL: 03

SUB CODE: 00 NR REF Sov: 004 OTHER: 004

Card 2/5

ACCESSION NO: A2300043

5/0170/63/006/005/0003/0006

AUTHOR: Vargaftik, N. N.; Zaytseva, L. S.

TITLE: Heat conductivity of deuterium in the gas phase

SOURCE: Inzhenerno-fizicheskiy zhurnal, v. 6, no. 5, 1963, 3-6

TOPIC TAGS: Deuterium; heat conductivity

ABSTRACT: The heat conductivity of D sub 2 O and H sub 2 O vapor was measured by the hot-wire method, using the apparatus shown in Fig. 1 of the Enclosure. Experimental curves (Fig. 2) have been obtained for a pressure  $\sim 9.8 \times 10^{-4}$  mm sup -2 and temperatures ranging from 100 to 500°C. The experiments have shown that the ratio of the heat conductivity of the two isotopes is a function of temperature (Fig. 3). A theoretical explanation of the results is offered in terms of statistical mechanics. Orig. art. lang: 5 equations, 4 figures, 2 tables.

Card 1/5

ZAYTSEVA, L.S., kand.tekhn.nauk

Experimental investigation of the thermal conductivity of the vapors of gasoline B-70, kerosene T-1, fuel T-5, and heptane.  
Trudy MAI no.132:79-93 '61.

(MIRA 14:7)

(Gases--Thermal properties)

Experimental investigation of

2/735/61/000/132/007/012  
5030/E484

maximum error in the results is  $\pm 1.3\%$ . There are 7 figures, 10 tables and 10 references, 6 Soviet and 4 non-Soviet. The three references to English language publications read as follows:  
Ref.7: Hilsenrath A., Touloukian D., Trans. ASME, v.76, no.6, 1954;  
Ref.9: Scherratt G., Griffiths E., Phys. Mag. and Journal of Sciences, v.27, no.180, 1939;  
Ref.10: Vines R., Austr. J. Chem. v.6, no.1, 1953.

Card 3/3

W

27861

S/535/61/000/132/007/012  
 Experimental investigation of ...  
 E030/E484

pressure region (around 400 mm Hg for convenience) and then assumed to be correct at the required pressure, to a maximum inaccuracy of 0.2%. To resolve the difficulties of nonhomogeneous chemical composition, the fuels were fractionated into three cuts for gasoline and four cuts for the other fuels, and data found for each cut. The mean values of conductivities may then be represented by the equation proposed by N.B.Vargaftik

$$\lambda = \lambda_0 (T/T_0)^n$$

where  $T_0$  is 273°K, and for gasoline,  $n = 1.95$  and  $\lambda_0$  is  $91 \times 10^{-4}$  kcal/metre hour °C, and for kerosene,  $n = 2.05$  and  $\lambda_0$  is  $74 \times 10^{-4}$  kcal/metre hour °C.  $\lambda$  can further be related to the boiling point  $t_k$  of a hydrocarbon by the formula:

$$10^4 \lambda_0 = 114 - 0.224 t_k$$

The check value for n-heptane was  $274 \times 10^{-4}$  kcal/metre hour °C, compared with the value of Moser (1913) of  $256 \times 10^{-4}$ . The Card 2/3

W

11.0100

27001

9/555/61/000/132/007/012  
E030/E484

AUTHOR: Zaytseva, L.S., Candidate of Technical Sciences

TITLE: Experimental investigation of the thermal conductivity of the vapours of gasoline B-70 (B-70), kerosene T-1, fuel T-5 and heptane

SOURCE: Moscow. Aviationsionnyy institut. Trudy. no.132.1961.79-93. Teplofizicheskiye svoystva nekotorykh aviatsionnykh topiliv v zhidknom i gazoobraznom sostoyanii,

TEXT: A heated wire apparatus was calibrated with air over the temperature range 0 to 500°C and used for the vapours of the above substances. The calibration agreed excellently with the best accepted values for air. The measuring tube was 155.5 mm long, made of glass with 3.98 mm id and 5.98 mm od; to minimize end heating corrections, the measuring wire was small, with 0.1 mm diameter, made of platinum, which served both as heater and resistance thermometer. To estimate the effects of the boundary layers as heat barriers, the experiments were carried out at pressures from 217 to 546 mm Hg for gasoline and, as expected, there was negligible variation in the thermal conductivity. All subsequent experiments were therefore conducted at a standard

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Classification: TS-FT-10 5955

squeous solution of caustic soda and in alkaline solution of potassium manganate, both solutions giving good results. Under the microscope using ultraviolet light, the W and Ti phases were clearly visible and readily distinguished from ferrite. Microhardness measurements made on the Ti-W-C alloy with an aqueous solution of oxalic acid gave the following values: Ti carbide, 1000; TiC, 1000; Ti to light brown under the microscope; carbide, brown and Cr, via UPM. Molybdenum carbide was brown and the molybdenum and ferrite areas remain light. The intermetallic phase in W prior to heating, tungsten carbide black and ferrite light green. Orig. art. has: 8% Mo, 10% Ti, 10% Cr.

Classification: Two-year delay per DIA memo dated 10/20/64 (Zerodur and Vycor glass)

Classification: TS-FT-10

BA/C/100

SUB COPY: MM

Classification: TS-FT-10

C-2/H/R-100

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001964100010-6

1	SILVER: investigation of the structure of iron-chromium-carbon and iron-molybdenum-carbon alloys by method of colorimetry with ultraviolet rays							
2	SOURCE: <i>Zhurnal poligraficheskogo instituta</i> , 7 (1964), no. 234, Metallovedenie i metallofizika, 14-25.							
3	REAGENTS: <i>Ultraviolet spectrophotometer</i> (ultraviolet radiation, concentration of solution, dilution, reagents for soda-boracium, potassium-manganate, etc.).							
4	APPARATUS: <i>Colorimeter</i> (ultraviolet radiation, concentration of solution, dilution, reagents for soda-boracium, potassium-manganate, etc.).							
5	METHOD: The authors have developed a method for the determination of chromium in the presence of cobalt (bright characteristic colors of the reaction products) and molybdenum (no reaction). Furthermore, the method can be used for the determination of other trivalent chlorides of Cr, Ti, V, Mn, Fe, Ni, Cu, Zn, Co, Cr <sup>2+</sup> and Cr <sup>3+</sup> in the presence of each other. The authors determined Cr and Cr <sup>2+</sup> in Mo-Cr and Cr-Mo-Cr alloys by titration with an							
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of new factors of the physics of an alloy. The authors propose a method of synthesis of single-phase intermetallic compounds, using various methods of preparation of a mixture and subsequently combining the

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of the values of the phases of an alloy, the corresponding specimens, using V-A-  
S-G-I and subsequently combining the  
values of the phases.

and either titanium or niobium in the chain. The colors of the individual layers of the intercalated manganese oxide are not clearly defined, but the following the structure of the clay may be clearly marked:

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Aggregates and Alloys

S/361/14/CB/234/3001/0017

Author: V. N. Kabanov, E. P. Fedorova, T. F.

Journal: Izvestiya Akademii Nauk SSSR, Seriya Metallovedeniya, No. 1, 1964, p. 107  
Title: Structure of Ti-Ni-C and Fe-Ni-C Alloys by method of colorimetry

Source: Izvestiya Akademii Nauk SSSR, Institute of Metals, no. 234, 1964. Metallurgicheskaya Literatura, No. 1, 1965.

Abstract: X-ray, titanium carbon, cobalt, colorimetry, ultraviolet radiation, infrared analysis, potassium manganate, chromium phosphate, intercalation, precipitation, microscopy, phase, microscope investigation.

Summary: The method of colorimetry makes a qualitative

TORBOCHKINA, L.I.; DORMIDOSHINA, T.A.; ZAYTSEVA, L.P.

Carbohydrate metabolism in cleandomycin-producing *Actinomyces antibioticus*. Mikrobiologija 33 no.1:162-166 Ja-F '64.  
(MIRA 17:9)

l. Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov  
(VNIIA).

Category : USSR/Solid State Physics - Morphology of Crystals. E-7  
Crystallization.

Abs Jour : Rof Zhur - Fizika, No 3, 1957, No 669

made of the effect of the speed of cooling of the alloy on the crystallization process. The cooling curves were recorded at various speeds with the Kurnakov instrument. The experimental curve of the change in the amount of liquid phase as a function of the hardening temperature lies below the theoretical curve, showing that the lever rule cannot be applied to the case of rapid cooling of alloys. The crystallization process of a solid solution under ordinary cooling conditions does not correspond to the scheme given by the equilibrium diagram. A new scheme is proposed for the change of composition of solid-solution crystals and of the liquid phase under unbalanced conditions. The author also believes that the dendrite structure of the solid solution results not only from dendrite liquation, but also from the crystallographic difference between the initial and final sections.

Card : 2/2

ZAYTSEVA, L.P.

Category : USSR/Solid State Physics - Morphology of Crystals. E-7  
Crystallization.

Abs Jour : Ref Zhur - Fizika, No 3, 1957, No 669?

Author : Zeytseva, L.P.

Title : Investigation of the process of Crystallization of a Solid  
Solution

Orig Pub : Tr. Leningr. politekhn. in-ta, 1955, No 180, 13-31

Abstract : The process of crystallization of a solid solution under  
actual cooling conditions was investigated in 14 alloys of  
the Cu-Si, Cu-Si-Mn, Al-Zn, and Sb-Bi systems. The ratio  
of the liquid and solid phases during the crystallization  
process was fixed for each given instant by hardening, after  
which a metallographic investigation was made of the struc-  
ture of the alloys on microsections, accompanied by measure-  
ment of the microhardness. The investigation of the change  
of concentration of the separating initial dendrites and of  
the parent solution was also carried out during the crystal-  
lization process with the aid of an ultraviolet color micro-  
scope and with the aid of local spectrography. A study was

Card : 1/2

of the liquid and solid phases during the crystallization process was fixed for each given instant by hardening, after which a metallographic investigation was made of the structure of the alloys on microsections, accompanied by measurement of the microhardness. The investigation of the change of concentration of the separating initial dendrites and of the parent solution was also carried out during the crystallization process with the aid of an ultraviolet color microscope and with the aid of local spectrography. A study was

Card : 1/2

source: Kof Zhur - Fizika, No 3, 1957, No 6699

made of the effect of the speed of cooling of the alloy on the crystallization process. The cooling curves were recorded at various speeds with the Kurnakov instrument. The experimental curve of the change in the amount of liquid phase as a function of the hardening temperature lies below the theoretical curve, showing that the lever rule cannot be applied to the case of rapid cooling of alloys. The crystallization process of a solid solution under ordinary cooling conditions does not correspond to the scheme given by the equilibrium diagram. A new scheme is proposed for the change of composition of solid-solution crystals and of the liquid phase under unbalanced conditions. The author also believes that the dendrite structure of the solid solution results not only from dendrite liquation, but also from the crystallographic difference between the initial and final sections.

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